

## Flash smelting

Direct smelting is now favored, e.g. using the following smelting technologies: [flash smelting](#)

([Finnish](#): *Liekkisulatus*) is a [smelting](#) process for [sulfur](#)-containing ores including [chalcopyrite](#). The process was developed by [Outokumpu](#) in [Finland](#) and first applied at the [Harjavalta](#) plant in 1949 for smelting copper ore. It has also been adapted for [nickel](#) and [lead](#) production.

A second flash smelting system was developed by the International Nickel Company ('[INCO](#)') and has a different concentrate feed design compared to the Outokumpu flash furnace. The **Inco flash furnace** has end-wall concentrate injection burners and a central waste gas off-take, while the **Outokumpu flash furnace** has a water-cooled reaction shaft at one end of the vessel and a waste gas off-take at the other end. While the INCO flash furnace at Sudbury was the first commercial use of oxygen flash smelting, fewer smelters use the INCO flash furnace than the Outokumpu flash furnace.

Flash smelting with oxygen-enriched air (the 'reaction gas') makes use of the energy contained in the concentrate to supply most of the energy required by the furnaces. The concentrate must be dried before it is injected into the furnaces and, in the case of the Outokumpu process, some of the furnaces use an optional heater to warm the reaction gas typically to 100–450 °C.

The reactions in the flash smelting furnaces produce copper [matte](#), iron oxides and [sulfur dioxide](#). The reacted particles

fall into a bath at the bottom of the furnace, where the iron oxides react with [fluxes](#), such as [silica](#) and [limestone](#), to form a [slag](#).

In most cases, the slag can be discarded, perhaps after some cleaning, and the matte is further treated in converters to produce blister copper. In some cases where the flash furnaces are fed with concentrate containing a sufficiently high copper content, the concentrate is converted directly to blister in a single Outokumpu furnace and further converting is unnecessary.

The sulfur dioxide produced by flash smelting is typically captured in a [sulfuric acid plant](#), removing the major environmental effect of smelting.

[Outotec](#), formerly the technology division of Outokumpu, now holds Outokumpu's patents to the technology and licenses it worldwide.

INCO was acquired by Brazil's [Vale](#) in 2006.

**In [flash smelting](#), the concentrate is dispersed in an air or oxygen stream and the smelting reactions are largely completed while the mineral particles are still in flight. The reacted particles then settle in a bath at the bottom of the furnace, where they behave as does calcine in a reverberatory furnace. A slag layer forms on top of the matte layer, and they can separately be tapped from the furnace.**